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**Warm-up**

**Problem 1** *Once Mary ate a half of the peaches from a can, the level of the syrup decreased by one third. How much would decrease the syrup level from the new one, if Mary eats a half of the peaches left?*

## Powers and Logarithms

Let  $b > 0$ . The following are the fundamental laws of exponential functions.

$$b^x \times b^y = b^{x+y} \quad (1)$$

$$b^0 = 1 \quad (2)$$

$$b^{-x} = \frac{1}{b^x} \quad (3)$$

$$\frac{b^x}{b^y} = b^{x-y} \quad (4)$$

$$(b^x)^y = b^{xy} \quad (5)$$

$$\sqrt[n]{b} = b^{\frac{1}{n}} \quad (6)$$

$$\sqrt[n]{b^m} = \left(\sqrt[n]{b}\right)^m = b^{\frac{m}{n}} \quad (7)$$

**Problem 2** Use the laws of the exponents to represent the following as powers of two.

- $2^5 \times 8^2 =$

- $\frac{2^5 \times 4^3}{16^2} =$

**Problem 3** Use the laws of the exponents to simplify the following as much as you can.

- $(a^x)^2 \times a^y \times a^{-2x} =$

- $\frac{(a^{-x})^{-y} \times a^3}{a^{xy-3}} =$

**Problem 4** Derive property 2 of exponential functions from property 1.

**Problem 5** *Derive property 3 from properties 1 and 2.*

**Problem 6** *Derive property 4 from properties 1 and 3.*

**Problem 7** *Derive property 6 from property 5.*

**Problem 8** *Derive property 7 from properties 5 and 6.*

**Problem 9** *To what power should one raise 10 to get 10,000,000?*

**Problem 10** *To what power should one raise 100 to get 10,000,000?*

**Problem 11** *To what power should one raise 0.5 to get 256?*

**Problem 12** *To what power should one raise  $1/9$  to get 3?*

**Problem 13** *Solve the following equation.*

$$3^{12x} = \left(\frac{1}{9}\right)^3$$



## Properties of logarithms

Since logarithmic functions are inverses of the exponential ones, properties of the logarithms mirror the formulas 1 - 5.

$$b^x \times b^y = b^{x+y} \qquad \log_b(pq) = \log_b p + \log_b q$$

$$b^0 = 1 \qquad \log_b 1 = 0$$

$$b^{-x} = \frac{1}{b^x} \qquad \log_b \frac{1}{p} = -\log_b p$$

$$\frac{b^x}{b^y} = b^{x-y} \qquad \log_b \left( \frac{p}{q} \right) = \log_b p - \log_b q$$

$$(b^x)^y = b^{xy} \qquad \log_b p^y = y \log_b p$$

**Problem 16** *Use 1 to prove the following formula.*

$$\log_b(pq) = \log_b p + \log_b q \qquad (8)$$

**Problem 17** Use 2 to prove the following formula.

$$\log_b 1 = 0. \quad (9)$$

**Problem 18** Use 3 to prove the following formula.

$$\log_b \left( \frac{1}{p} \right) = -\log_b p \quad (10)$$

**Problem 19** Prove the following formula.

$$\log_b \left( \frac{p}{q} \right) = \log_b p - \log_b q \quad (11)$$



**Problem 20** Use 5 to prove the following formula.

$$\log_b p^y = y \log_b p \quad (12)$$

**Problem 21** How is  $\log_b \sqrt{x}$  related to  $\log_b x$ ?

**Problem 22** How is  $\log_b \sqrt[n]{x}$  related to  $\log_b x$ ?

**Problem 23** Find the following values.

- $\log_{\sqrt{3}} 81 =$

- $\log_6 3 + \log_6 2 =$

- $\log_2 2\sqrt{2} =$

- $2\log_4 2 =$

**Problem 24** Solve the following equation.

$$4^{\sqrt{x+1}} = 64 \times 2^{\sqrt{x+1}}$$

## Carbon Dating

$^{14}\text{C}$  is a radioactive isotope of carbon with *half-life time*  $t_{1/2} = 5,700$  years. This means that if the amount of  $^{14}\text{C}$  you have right now is  $A_0 = 10$  grams, then 5,700 years later the amount left will be a half of that,  $A(5,700) = 5$  grams.

### Problem 25

- *What amount of the isotope would be left  $2 \times 5,700 = 11,400$  years from now?*

- $3 \times 5,700 = 17,100$  years from now?

- $4 \times 5,700 = 22,800$  years from now?

- $n \times 5,700$  years from now?

**Problem 26** *If the original amount of  $^{14}\text{C}$  was  $A_0$ , how much would be left  $T$  years from now? The following formula should help.*

$$T = \frac{T}{t_{1/2}} \times t_{1/2} \quad (13)$$

$^{14}\text{C}$  is produced by cosmic rays bombarding nitrogen atoms in the Earth's atmosphere. The resulting radioactive carbon combines with atmospheric oxygen to form radioactive carbon dioxide gas  $\text{CO}_2$ . The latter is digested by plants in the process of [photosynthesis](#). It is passed further down the food chain from plants to plant-eating animals to carnivores. A living plant, or animal, has a stable level of  $^{14}\text{C}$  in its body. Once the plant, or animal, dies, it stops exchanging the isotope with the environment. The level of  $^{14}\text{C}$  starts decreasing according to the equation you have found in Problem 26. The idea to use the process of the  $^{14}\text{C}$  decay for dating biological samples belongs to [Willard Libby](#) who received the Nobel Prize for his work in 1960.

**Problem 27** *A petrified redwood log has 1% of the original  $^{14}\text{C}$  amount. How old is the sample?*